

$$\gamma = \frac{1}{\sqrt{1 - \frac{v^2}{c^2}}}$$

Time dilation $\Delta t = \gamma \Delta t_0$

Length Contraction $l = \frac{l_0}{\gamma}$

Lorentz Transformations

$$x' = \gamma(x - ut) \quad y' = y \quad z' = z$$

$$t' = \gamma(t - \frac{ux}{c^2})$$

$$v'_x = \frac{v_x - u}{(1 - \frac{u v_x}{c^2})}$$

$$v_x = \frac{v'_x + u}{(1 + \frac{u v'_x}{c^2})}$$

$$\text{Doppler shift } f = \sqrt{\frac{c+u}{c-u}} f_0$$

Relativistic energy and momentum

$$\vec{p} = \gamma m \vec{v}$$

$$K = (\gamma - 1)mc^2$$

$$E = K + mc^2 = \gamma mc^2$$

$$E^2 = (mc^2)^2 + (pc)^2$$

$$\text{Photon Energy } E = hf = \frac{hc}{\lambda}$$

$$\text{Photon momentum } E = pc$$

$$\text{Photoelectric Effect } eV_0 = hf - \phi$$

$$\text{Atomic Line Spectra } hf = E_i - E_f$$

$$\text{X-ray production } E_{max} = eV_{AC} = \frac{hc}{\lambda_{min}}$$

$$\text{Compton scattering: } \lambda' - \lambda = \frac{h}{mc}(1 - \cos \phi)$$

$$\text{Stefan-Boltzmann law } I = \sigma T^4$$

$$\text{Wien Displacement law } \lambda_m T = 2.9 \cdot 10^{-3} \text{ m} \cdot \text{K}$$

$$\text{Planck Radiation Law } I(\lambda) = \frac{2\pi hc^2}{\lambda^5(e^{\frac{hc}{\lambda kT}} - 1)}$$

$$\text{de Broglie Waves } \lambda = \frac{h}{p} = \frac{h}{\sqrt{2meV_{ba}}}$$

Heisenberg's uncertainty principle

$$\Delta x \Delta p_x \geq \frac{\hbar}{2} \quad \Delta E \Delta t \geq \frac{\hbar}{2}$$

Stationary state wave function

$$\Psi(x, y, z, t) = \psi(x, y, z) e^{iEt/\hbar}$$

Schrödinger equation

$$(-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + U(x))\psi(x) = E\psi(x)$$

Particle in a box

$$E_n = \frac{p_n^2}{2m} = \frac{n^2 \pi^2 \hbar^2}{2m L^2}$$

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin \frac{n\pi x}{L}$$

$n = 1, 2, 3, \dots$

Wave function normalization

$$\int_{-\infty}^{\infty} |\psi(x)|^2 dx = 1$$

$$P(r)dr = |\psi(r)|^2 dV = |\psi(r)|^2 4\pi r^2 dr$$

Atomic Energy Levels

$$E_n = -\frac{1}{(4\pi\epsilon_0)^2} \frac{me^4}{2n^2\hbar^2} = -\frac{13.6eV}{n^2}$$

$$E_n = -\frac{Z_{eff}^2}{n^2} 13.6eV$$

$$a = \frac{4\pi\epsilon_0\hbar^2}{me^2} = 5.3 \cdot 10^{-11} \text{ m}$$

Moseley's law $f = (2.48 \cdot 10^{15} Hz)(Z - 1)^2$

Angular momentum

$$L = \sqrt{l(l+1)}\hbar \quad L_z = m_l\hbar$$

Spin

$$S = \sqrt{s(s+1)}\hbar = \sqrt{\frac{1}{2}(\frac{1}{2}+1)}\hbar \quad S_z = \pm \frac{1}{2}\hbar$$

Zeeman effect

$$U = -\mu_z B = m_l \frac{e\hbar}{2m} B = m_l \mu_B B$$

Exclusion Principle, Quantum numbers

$$n \geq 1$$

$$0 \leq l \leq n-1$$

$$|m_l| \leq l$$

$$m_s \pm \frac{1}{2}$$

Molecules

$$E_l = l(l+1) \frac{\hbar^2}{2I} \quad l = 0, 1, 2, 3, \dots$$

$$I = m_r r_0^2 \quad m_r = \frac{m_1 m_2}{m_1 + m_2}$$

$$E_n = (n + \frac{1}{2})\omega\hbar = (n + \frac{1}{2})\sqrt{\frac{k'}{m_r}}\hbar$$

$$n = 0, 1, 2, 3, \dots$$

Free Electron Model

$$g(E) = \frac{(2m)^{\frac{3}{2}} V}{2\pi^2 \hbar^3} E^{\frac{1}{2}}$$

$$f(E) = \frac{1}{e^{\frac{E-E_F}{kT}} + 1}$$

p-n junction diode

$$I = I_S(e^{\frac{eV}{kT}} - 1)$$

Nuclear Radius

$$R = R_0 A^{\frac{1}{3}} \quad R_0 = 1.2 \cdot 10^{-15} \text{ m}$$

Nuclear Binding Energy

$$E_B = (ZM_H + Nm_n - Z_A M)c^2$$

Radioactive Decay

$$N(t) = N_0 e^{-\lambda t}$$

Speed of light

$$c = 3 \cdot 10^8 \text{ m/s}$$

Elementary Charge

$$e = 1.6 \cdot 10^{-19} \text{ C}$$

Gravitational constant

$$G = 6.67 \cdot 10^{-11} \text{ N} \cdot \text{m}^2/\text{kg}^2$$

Planck's constant

$$h = 6.626 \cdot 10^{-34} \text{ Js}$$

Boltzmann's constant

$$k = 1.38 \cdot 10^{-23} \text{ J/K}$$

Electron rest energy

$$m_e c^2 = 511 \text{ keV}$$

Proton rest energy

$$m_p c^2 = 938.3 \text{ MeV}$$

Neutron rest energy

$$m_n c^2 = 939.6 \text{ MeV}$$